Sesquiterpenoids from *Dysoxylum amooroides* **Stem Bark: Isolation, Structure Determination, and Cytotoxicity Against MCF-7 Breast Cancer Cells**

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Keywords: cytotoxic activity; Dysoxylum amooroides; MCF-7; sesquiterpenoids

■ **INTRODUCTION**

Sesquiterpenoids are widely distributed in nature, especially within higher plants, and more than 10,000 sesquiterpenoids have been isolated. Sesquiterpenoids are natural organic compounds that belong to the terpenoid group. They comprise three isoprene units (C_5) , generally containing 15 carbons and 24 hydrogens per molecule $(C_{15}H_{24})$. Since sesquiterpenoids are formed by the farnesyl pyrophosphate, a precursor that is produced by 2 building blocks, isopentenyl diphosphate and dimethylallyl diphosphate from 3 units of acetyl-CoA via the mevalonate pathway, these compounds exhibit abundant structural diversity, with many found in cyclic

forms by the existence of 3 double bonds and more flexible carbon chains [1-3]. Sesquiterpenoids are often used as fragrance ingredients because they are the main constituents of essential oils. The biological potency of sesquiterpenoid compounds has been extensively reported, including antibacterial, antifungal, antimalarial, anti-inflammatory, anticarcinogenic, antitumor, cytotoxic, and immunomodulatory based on toll-like receptor 4 (TLR4) [4-19].

Meliaceae belongs to the order Sapindales and consists of tropical plants renowned for their highquality wood and aromatic stems. This family encompasses 58 genera and approximately 740 species

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[15]. Sesquiterpenoids are distributed in various genera in the Meliaceae family, namely Aglaia [20], Dysoxylum [21], Chisocheton [22], Guarea [23], Trichilia [24], and Lansium [25]. The Dysoxylum is distributed worldwide, predominantly in tropical and subtropical regions such as China, India, Malaysia, Northeast Australia, and various Southeast Asian countries. This genus has a characteristic tree height of ±36 m and comprises around 200 species of these compounds belonging to sesquiterpenoid [26], sesquiterpenoid dimers [27], diterpenoids [28-29], triterpenoids [30-33], limonoids [34-37], and macrolides [38]. Plants of the genus Dysoxylum are widely utilized in traditional medicine to treat various ailments. For instance, the leaves of D. richii are brewed into tea to alleviate pain and are often used as furniture because the wood has a high quality. It is also used in traditional medicine, such as D. malabaricum, known to cure rheumatism, and its oil is used as an eye and ear medicine [39]. Approximately 53 sesquiterpenoids have been successfully isolated from this genus [15]. 10β-hydroxy-4α,4β-dimethyl-5αH,7αH-eudesm-3-one, isolated from the stem bark of D. parasiticum, exhibits significant cytotoxic activity against MCF-7 cancer cells, with an IC_{50} value of 27.39 μM [40]. These results inspire us to explore further sesquiterpenoids from the Dysoxylum genus and their action in cytotoxic activity against breast cancer MCF-7 cells.

One species of the Dysoxylum that has not been extensively explored for its chemical compounds is D. amooroides. The investigation of the stem bark of D. amooroides yielded 3 compounds, including 1 previously described guaiane-type sesquiterpenoids guaianediol (**1**) and alismol (**2**), as well as 1 previously described aromadendrane-type sesquiterpenoid spathulenol (**3**). In this work, their isolation and structure elucidation will be described thoroughly to understand further the phytochemical study of sesquiterpenoids **1**–**3** from D. amooroides. A modest activity against breast cancer MCF-7 cells was obtained by inhibiting **3**, which was more potent than its reference drug cisplatin. This study also provides a brief structure-activity relationship based on the similarity of the basic framework in the biogenesis pathway that might be useful for developing drug discovery on the synthetic approach.

■ **EXPERIMENTAL SECTION**

Materials

The plant of D. amooroides Miq. stem bark was collected from the Pangandaran Nature Reserve, Pangandaran, West Java province, Indonesia, in February 2021. The plant was identified and made by Mr. Joko Kusmoro and was deposited at the Plant Taxonomy Laboratory, Department of Biology, Universitas Padjadjaran (specimen No. 42/HB/07/2021).

Instrumentation

The instruments used are lab glassware and in the separation process vacuum liquid chromatography and common column chromatography (CC) were performed on silica gel 60 (Merck KGaA, Darmstadt, Germany, 70–230 and 230–400 mesh) and octadecyl silane (ODS, Chromatorex C₁₈ DM1020T, Fuji Sylisia Chemical Ltd., Japan, 100–200 mesh). The isolation guidance was utilized in a thin layer chromatography (TLC) using silica gel 60 F₂₅₄ (Merck KGaA, Darmstadt, Germany) plates of the normal phase and the reverse one using reverse phase RP-18 F_{254S} plates (Merck KGaA, Darmstadt, Germany). The TLC plates were analyzed under UV light at 254 and 365 nm wavelengths and sprayed with 10% H2SO4 in EtOH. Infrared spectra were measured on a Perkin-Elmer Spectrum 100 with KBr disk (Waltham, Massachusetts, USA). Mass spectra were measured with a waters XEVO HR-TOF-ESI-MS (Milford, MA) complemented with ESI+ mode. The NMR spectra were recorded with internal standard as a TMS on a Bruker Ascend spectrometer, involving ¹H at 700 MHz, ¹³C at 175 MHz, DEPT at 135 MHz, ¹H-¹H COSY, HMBC, HSQC, and NOESY. A cytotoxic assay was conducted using a microplate reader Infinite M200 (TECAN, Switzerland) at λ_{max} 570 nm, a 96-well plate (Thermo Fisher Scientific Inc., USA), and an incubator.

Procedure

Extraction and isolation

The stem bark of *D. amooroides* was prepared in a preliminary step by drying and powdering (2.8 kg) and was then macerated with EtOH at room temperature (3×12) L, 24 h each). The macerates were then evaporated, and the solvent under an evaporator decompression was obtained as a crude extract of EtOH (491.3 g), eluted in H_2O , and partitioned sequentially based on polarity with n -hexane and EtOAc. The solvent was removed by an evaporator to afford *n*-hexane (30 g) and EtOAc (110.6 g) extracts.

The concentrated *n*-hexane extract of 30 g was separated by vacuum liquid chromatography using a polar stationary phase (on silica gel) with the eluent n hexane-EtOAc (100:0–0:100, 10% v/v) to obtain 9 fractions (Fr. A–I). Furthermore, 2.63 g of Fr. C was further separated by CC with *n*-hexane:EtOAc $(60:1)$ as eluent to afford 11 subfractions (C1–C11). Using CC with n-hexane:EtOAc (60:1) as eluent, subfraction C5 (420 mg) was separated into 5 subfractions (C5a–C5e). Subfraction C5b (22.4 mg) was purified by ODS CC (acetonitrile: H_2O , 7:3) to afford compound 2 (11.3 mg). Compound **3** was purified by flash ODS CC (acetonitrile: H_2O , 7:3) from subfraction C5c, weighing 17.3 mg.

The concentrated EtOAc extract of 110.6 g was separated by vacuum liquid chromatography of silica gel with n-hexane:EtOAc and EtOAc:MeOH (100:0–0:100, 10% v/v) gradient elution to yield 5 fractions (Fr. A-E). Furthermore, Fr. B weighing 680 mg was further separated with CC with n -hexane:EtOAc (8.5:1.5) to get 11 fractions (B1–B11). Then, subfraction B6 (32.7 mg) was separated by CC on ODS with MeOH: H_2O (7:3) eluent to produce 5 fractions (B6a–B6e). Then, purified fraction B6b by flash ODS CC (MeOH:H2O, 4:6) to obtain compound **1** (4.1 mg).

Compound 1. Colorless oil. IR (KBr) v_{max} 3300, 2900, 1600, 1383, 1384, 1150 cm⁻¹. ¹H-NMR (CDCl₃, 700 MHz) is presented in Table S1 and ¹³C-NMR (CDCl₃, 175 MHz) data is presented in Table 1. HR-TOF-ESI-MS m/z 239.2013 [M+H]⁺ (calc. for C₁₅H₂₆O₂, *m/z* 239.2011).

Compound 2. Colorless oil. IR (KBr) ν_{max} 3372, 2958, 2862, 1638, 1461, 1379 cm⁻¹. ¹H-NMR (CDCl₃, 700 MHz) is presented in Table S1 and 13C-NMR (CDCl3, 175 MHz) data is presented in Table 2. HR-TOF-ESI-MS m/z 259.1453 [M+K]⁺ (calc. for C₁₅H₂₄O, m/z 259.1464).

Compound 3. Colorless oil. IR (KBr) ν_{max} 3396, 2926, 2856, 1636, 1457, 1376 cm⁻¹. ¹H-NMR (CDCl₃, 700 MHz) is presented in Table S1 and 13C-NMR (CDCl3, 175 MHz) data is presented in Table 3. HR-TOF-ESI-MS m/z 259.1453 [M+K]⁺ (calc. for C₁₅H₂₄O, m/z 259.1464).

Cytotoxic activity assay

All the isolated compounds **1-3** were evaluated for cytotoxicity against human MCF-7 breast cancer cells using the resazurin method by measuring cell viability with PrestoBlue® reagent. The cells were cultured in RPMI-1640 medium, consisting of 50 μL/50 mL of antibiotic (1% penicillin) and 10% fetal bovine serum. The cells were cultured in 96-well plates and incubated for 24 h at 37 °C in 5% $CO₂$ gas. RPMI media was disposed of, and then the sample media and positive control cisplatin in DMSO with required concentrations (500.00; 250.00; 125.00; 62.50; 31.25; 15.63; 7.81; 3.91 μg/mL) were added, respectively. Cells were incubated for 48 h after being treated with samples and positive controls. The media containing the sample was discarded and then incubated for 2 h with PrestoBlue® reagent until a color change occurred. The samples were measured using a multimode reader to determine their absorbance at 570 nm, and absorbance was converted into cell viability values to determine the IC₅₀ value of each compound [34].

■ **RESULTS AND DISCUSSION**

The *n*-hexane and EtOAc extracts were repeatedly separated and purified by normal and reversed-phase CC [15,20-21,27], producing 3 sesquiterpenoids **1**-**3** (Fig. 1). Compound **1** was isolated as a colorless oil with a yield of 4.1 mg, isolated from EtOAc extract, with a molecular formula of $C_{15}H_{26}O_2$ established by HRTOF-ESI-MS m/z 239.2013 (calc. for [M+H]⁺ m/z 239.2011),

Fig 1. Structure of sesquiterpenoid compounds **1**-**3**

which represents 3 unsaturation degrees. The IR spectrum (Fig. S2) showed absorption bands of hydroxyl group (3300 cm⁻¹), CH sp³ group (2900 cm⁻¹), C=C group (1600 cm−1), gem-dimethyl group (1383 and 1384 cm−1) and C–O group (1150 cm⁻¹). The ¹H-NMR data showed 2 tertiary methyl at δ_H (ppm) 1.21 (s, CH₃-14) and 1.27 (s, CH₃-15), 2 doublet signals of secondary methyl at δ_H 0.98 $(d, CH₃-13)$ and 0.99 $(d, CH₃-12)$, and 1 olefin proton at δ_H shift 5.48 (br.s, CH-6). Furthermore, ¹³C-NMR data (Table 1) supported with DEPT and HSQC, showed 15 carbons signals, including 4 methyl at δ_c (ppm) 21.1 (C-14), 21.3 (C-13), 21.4 (C-12) and 22.5 (C-15), 4 $sp³$ methylene at δ_C 21.5 (C-2), 25.1 (C-8), 80.3 (C-4) and 42.6

(C-9), 3 sp^3 methine at δ_c 37.3 (C-11), 50.3 (C-1) and 50.3 (C-5), 1 sp^2 methine at δ_c 121,3 (C-6), 2 oxygenated quaternary carbons present at δ _C 75.3 (C-10), 80.3 (C-4), and 1 sp² quaternary carbon at δ _C 149.6 (C-7). The above data suggested that compound **1** is a sesquiterpenoid compound with 3 degrees of unsaturation. The existence of one pair double bond (δ _C 121.3, 149.6) indicated that the remaining unsaturation in **1** was attributed to 2 rings (bicyclic system) of sesquiterpenoid.

The 2D-NMR analysis showed HMBC correlations of H-5 (δ_H 2.16) to C-4 (δ_C 80.3) and C-1 (δ_C 50.3), suggesting that **1** is a guaiane-type sesquiterpenoid. Two tertiary methyl by hydroxyl attachments at C-4 and C-10 was confirmed by HMBC correlations of CH_{3} -15 (δ_{H}) 1.27) to C-4 (δ _C 80.3) and C-3 (δ _C 40.5) and CH₃-14 (δ _H 1.21) to C-9 (δ_c 42.6), C-10 (δ_c 75.3) and C-1 (δ_c 50.3), respectively. The isopropyl moiety at C-7 as well as a pair double bond positioned at C-6/C-7 were assigned based on the observed correlations of CH₃-13 (δ _H 0.98)/CH₃-12 (δ_H 0.99) to C-11 (δ_C 37.3) and C-7 (δ_C 149.6) and H-1 (δ_H 1.88) to C-6 (δ_C 121.3). In addition, the ¹H-¹H COSY

Position of carbon	1	Guanaediol $[41]$ *	
	δ_H ppm (ΣH , mult., $J = Hz$)	δ _C (mult.)	
$\mathbf{1}$	1.88(1H, m)	50.3 (d)	50.7(d)
$\overline{2}$	1.64 (1H, m)	21.5(t)	21.5(t)
	1.77(1H, m)		
3	1.71 (1H, m)	40.5(t)	40.5(t)
	1.61 (1H, m)		
$\overline{4}$		80.3(s)	80.2(s)
5	2.16 (1H, m)	50.3 (d)	50.3 (d)
6	5.48 (1H, brs)	121.3 (d)	121.3 (d)
7		149.6 (s)	149.6 (s)
8	2.20(1H,m)	25.1(t)	25.1(t)
	1.92(1H, m)		
9	1.82 (1H, m)	42.6(t)	42.6(t)
	1.47(1H, m)		
10		75.3(s)	75.3(s)
11	2.25 (1H, m)	37.3 (d)	37.3 (d)
12	0.99 (3H, d , 5.5)	21.4(q)	21.4(q)
13	0.98 (3H, d , 5.5)	21.3(q)	21.3(q)
14	1.21(3H, s)	22.5(q)	22.5(q)
15	1.27(3H, s)	21.1(q)	21.1(q)

Table 1. 13C-NMR data for compound **1** and their related literature

*(CDCl3, 125 MHz)

cross-peaks of H-1/H-2/H-3, H-1/H-5/H-6, and H-9/H-8 convinced the planar structure of **1** by 2 adjacent vicinal proton systems. The fusion of five and seven-membered rings with 2 hydroxyls and one pair of olefinic bond groups in **1** was deduced. Since the guaiane compound in **1** shows several stereocenter carbons, the relative configuration is mandatorily required to be established. A NOESY experiment was then performed to observe the key cross-peaks between 2 protons from asymmetrical carbons in a particular space. In the H - H NOESY spectrum, the key cross-peaks between H-5 (α-oriented) to CH_3 -14 and H-2a indicated that CH_3 -14 and H-5 are α oriented. Conversely, the cross-peak of H-2b/CH3-15 suggested the β-orientation of CH_3 -15 and α -orientation of hydroxyl at C-4. A comparison of 1D-NMR data between **1** and the related literature showed a significant resemblance to the guaianediol compound. Hence, compound **1** was identified as a guaianediol like previous literature [41]. The MS, FTIR, ¹H-NMR, ¹³C-DEPT NMR, HSQC, HMBC, ¹H-¹H COSY, and NOESY spectra were presented in Fig S1-S8.

Compound **2** was isolated as a colorless oil of 11.3 mg from n -hexane extract. Its molecular formula as $C_{15}H_{24}O$ was determined by HRTOF-ESI-MS m/z 259.1453 (calc. for $[M+K]^+$ m/z 259.1464), which represents 4 degrees of unsaturation. The IR spectrum (Fig. S10) implied the existence of hydroxyl (3372 cm⁻¹), the aliphatic of C-H sp^3 (2958 and 2869 cm⁻¹), C=C sp^2 (1638 cm⁻¹), and gem-dimethyl (1461 and 1379 cm⁻¹) groups. 1 H-NMR data (Table 2) showed 3 methyl, including 2 secondary methyl at δ_H (ppm) 0.98 (3H, d, J $= 6.0$ Hz, CH₃-13), 0.99 (3H, d, J = 6.0 Hz, CH₃-12) and 1 tertiary at δ_H 1.25 (3H, s, CH₃-15). The presence of a sp² methylene at δ_H 4.73 and 4.76 (1H, s, H-14a/H-14b) and a methine sp^2 at δ_H 5.55 (1H, s, H-6) were also observed. The 13C-NMR data, with the aid of DEPT and HSQC spectrum, retrieved 15 carbons, which consists of 3 $s p³$ methyl at δ _C 21.2 (C-12), 21.5 (C-13), and 24.0 (C-15), 4 sp^3 methylene at δ_c 24.8 (C-2), 40.3 (C-3), 30.0 (C-8), and 37.1 (C-9), 1 sp^2 methylene at δ_c 106.5 (C-14), and 3 methine sp^3 at δ_C 47.3 (C-1), 55.1 (C-5), and 37.1 (C-11), 1 sp² methine at δ _C 121.5 (C-6), as well as 1 oxygenated

*(CDCl3, 100 MHz)

non-protonated carbon at δ_c 80.8 ppm (C-4). The presence of 2 non-protonated sp^2 carbons at δ_c 148.7 (C-7) and 154.1 (C-10) confirmed that **2** bears two pairs of double bonds. The above data showed that **2** was also a bicyclic guaiane-type sesquiterpenoid. The 2D-NMR analysis (Fig. 2 and 3) revealed that **2** resembled **1**, in which the methylene olefinic group replaced hydroxy methyl at C-10 in **1**. This assignment was proved by HMBC correlations from H-14 at δ_H 4.76 (1H, s) and 4.73 (1H, s) to C-10 (δ_c 154.1), C-9 (δ_c 37.1), and C-1 (δ_c 47.3). The relative configuration of **2** was also assigned as **1** by NOE correlations of CH3-15 to H-1 (β-oriented), and no observed cross-peaks between H-5 to β protons. Further comparison of 1D-NMR data between **2** and the related literature showed high similarity with the known compound alismol. Thus, compound **2** was identified as the same as in previous literature [42]. The MS, FTIR, ¹H-NMR, ¹³C-DEPT NMR, HMBC, ¹H-¹H COSY, and NOESY spectra were presented in Fig S9-S15.

Compound **3** was isolated, having the characteristics of colorless oil of 17.3 mg, isolated from n-hexane extract. Its molecular formula of $C_{15}H_{24}O$ was determined based on HRTOF-ESI-MS m/z 259.1453 (calc. for $[M+K]^+ m/z$ 259.1464), representing 4 unsaturation degrees. The IR spectrum (Fig. S17) suggested the presence of hydroxyl group (3396 cm⁻¹) with typical bandwidth, C-H sp³ (2926 and 2856 cm⁻¹), C=C sp² (1636 cm⁻¹), and gem-dimethyl (1457 and 1376 cm−1) groups. The 1 H-NMR data showed 3 methyl at δ_H 1.03 (CH₃-12), 1.04 (CH₃-13), and 1.28 (3H, s, CH₃-15), and a pair sp² methylene at δ_H 4.68 (1H, s, H-

14a) and 4.70 (1H, s , H-14b). Furthermore, the¹³C-NMR corroborated with DEPT data possessed 15 carbons resonances, involving 3 sp^3 methyl at δ_c 16.5 (C-12), 28.8 (C-13), and 26.8 (C-15), 4 sp^3 methylene at δ_c 26.7 (C-2) 41.9 (C-3) 24.4 (C-8) and 39.0 (C-9), 1 methylene olefinic at δ_C 106.4 ppm (C-14), 4 sp^3 methine at δ_C 53.5 (C-1), 54.9 (C-5), 30.0 (C-6), and 27.6 (C-7), 1 sp^3 nonprotonated carbon at δ_C 20.4 ppm (C-11), 1 oxygenated at δ_c 81.1 ppm (C-4), and 1 sp² non-protonated carbon at δ _C 153.6 (C-10). Together with the fact that no other olefinic bonds were present, compound **3** was then deduced to share a tricyclic sesquiterpenoid scaffold. The presence of a sp^3 quaternary carbon at δ_c 20.4 ppm (C-11) and the shielded protons at δ_H 0.44 (1H, *dd*, *J* = 9.2, 10.7 Hz, H-6), 0.71 (1H, m, H-7) firmly pointed out that **3** is an aromadendrane-type sesquiterpenoid by a cyclopropane moiety positioned at C-6/C-7/C-11. Based on the above assignment, compound **3** was deduced as an aromadendrane bearing 3 tertiary methyls with one attached to hydroxyl quaternary carbon and a pair of

Fig 2. Selected HMBC and ¹H-¹H COSY Correlation of compounds **1** and **2**

Fig 3. NOESY correlation of compounds **1** and **2**

olefinic bonds constituted by the $[-CH_2=C-]$ system. Further literacy to compound **3** with a known aromadendrane-type sesquiterpenoid spathulenol disclosed that their 1D-NMR was nearly identical (Table 3). Therefore, compound **3** was identified as 4β-hydroxy-10-en-aromadendran spathulenol [43]. The MS, FTIR, ¹H-NMR, and ¹³C-DEPT NMR spectra were presented in Fig S16-S19.

The activity of compounds **1**-**3** were evaluated against MCF-7 breast cancer cells by using the resazurin method with the positive control of cisplatin, as shown in Table 4. The results showed that all compounds had varying toxicity from inactive to moderate. The cytotoxicity compounds were classified on the basis of a previous literature report for pure compounds as highly active $(IC_{50} < 2 \mu M)$, moderately active $(IC_{50} < 10 \mu M)$, and inactive $(IC₅₀ > 100 μM)$ [44]. Compound 3 showed the strongest activity with an IC_{50} value of 15.2 μ M, followed by 2 with weak activity (an IC_{50} value of 82.1 μ M), while 1 was inactive with an IC₅₀ value of > 100 μM against MCF-7 cancer cells. The potent inhibition played by compound **3** in this study attracts us to verify its activity further. A previous study by Naini et al. [40] agreed that spathulenol (**3**) was more potent than cisplatin, with an IC_{50} value of 12.2 μ M. Furthermore, a significant enhancement activity of **2** compared to **1** in a guaiane-type revealed that $[-CH_2=C-]$ at $C-10/C-14$ might facilitate the response of **2**, while a pair olefinic bond [–CH=C–] at C-6/C-7 did not contribute against breast cancer cell's growth MCF-7. Beyond that, the formation of cyclopropane at C-6/C-7/C-11 yielded an aromadendrane-type in **3**, resulting in a superior activity more than 5-fold compared to **2** (Fig. 4). These findings sharpen the study of the discovery of sesquiterpenoids with biological activity as cytotoxic agents against cancer cells, especially the naturally occurring aromadendrane framework with considerable activity.

Table 4. Cytotoxicity of **1**-**3** against MCF-7 cell lines

Compound	$IC_{50}(\mu M)$		
Guaianediol (1)	>100		
Alismol (2)	82.1		
Spathulenol (3)	15.2		
Cisplatin (positive control)	53.0		

Position of carbon	3	spathulenol [43]*	
	δ_H ppm (ΣH , mult., $J = Hz$)	$\delta_{\rm C}$ (mult.)	
$\mathbf{1}$	1.31 (1H, m)	53.5(d)	53.4 (d)
$\overline{2}$	1.88 (1H, dd, 6.0, 12.0)	26.7(t)	
	1.64 (1H, dd, 6.0, 12.0)		26.7(t)
3	1.76 (1H, m), 1.54 (1H, m)	41.9(t)	41.7(t)
$\overline{4}$		81.1(s)	81.0(s)
5	1.31 (1H, m)	54.5 (d)	54.3 (d)
6	0.44 (1H, dd, 9.0, 10.7)	30.0(d)	29.9(d)
7	0.71 (1H, m)	27.6 (d)	27.5 (d)
8	1.96(2H, m)	24.4(t)	24.8(t)
9	2.41 (1H, dd, 6.0, 13.5)	39.0 (t)	
	2.04 (1H, dd, 6.0, 13.5)		38.9(t)
10		153.6(s)	153.4(s)
11		20.4(s)	20.3(s)
12	1.03 (3H, s)	16.4 (q)	16.3 (q)
13	1.04(3H, s)	28.8(q)	28.7(q)
14	4.68 (1H, s), 4.70 (1H, s)	106.4(t)	106.3(t)
15	1.28(3H, s)	26.8(q)	26.1(q)

Table 3. 13C-NMR data for compound **3** and its related literature

 $*(CDCl₃, 150 MHz)$

Since compounds **1**-**3** share the same 5/7-bicyclic fused ring skeleton with a diverse activity in inhibiting the proliferation of cancer cells, a brief exploration with those of previous works was prompted to gain more understanding of the existence of this group in the biological activity window. As mentioned, guaianediol (**1**) has not demonstrated significant activity against breast cancer cells. However, according to the previous work conducted by Li et al. [45], compound **1** showed potential antimicrobial activity, particularly against Staphylococcus aureus, with minimum inhibitory concentrations (MIC) recorded at 32 μg/mL. These findings suggest that guaianediol (**1**) may be helpful in representing a potent candidate for further development as an antibacterial agent.

Regarding its cytotoxic effects, alismol (**2**) provides a revealed inhibition compared to **1** by forming a terminal olefinic bond to replace the hydroxyl methyl at C-10/C-14. These results were supported by a previous work that **2** has a notable cytotoxicity against HT-29, A-549, and A-2058 cancer cells, with the IC_{50} consecutively exhibiting values of 29.8 ± 3.0 , 25.4 ± 3.6 , and 26.2 ± 2.9 μ M [46]. Regardless of the cytotoxicity of alismol (**2**), which was categorized as moderately active against several human cancer cells [44], its potency as a naturally cytotoxic compound is worth considering. Moreover, alismol (**2**) was reported to potentially reduce lung inflammation, which mechanistically activates the nuclear factor erythroid 2-related Ffctor 2 (Nrf2) pathway and induces Nrf2-dependent gene expression without inhibiting NF- κB and also demonstrated as a potent compound in the treatment of respiratory disorders [47].

In addition to the strong activity against human breast MCF-7 cancer cells, spathulenol (**3**) was found to have modest activity against HT-29, A-549, and A-2058 cancer cells, with its IC₅₀ values of 25.4 ± 4.6 , 22.9 ± 2.8 , and $21.6 \pm 2.0 \mu M$, respectively [46]. Although the previous findings disclosed lower cytotoxicity results than those against MCF-7 cells concluded in this work, spathulenol (**3**) deserves to be considered a potent cytotoxic agent with selective activity against several panel cancer cells. A versatile biologically active compound from nature in **3** was also proved by a strong antibacterial activity against Mycobacterium tuberculosis, with both MIC and minimum bacterial concentration (MBC) values of 6.25 μg/mL [48]. Thus, spathulenol (**3**) provided a broad-spectrum activity, which might be useful as a "gold-mine" compound for drug development.

Based on the current cytotoxicity results, structure-activity relationship analysis, and previous works, this information significantly contributes to the growing research on sesquiterpenoids. The potent biological properties of these natural sesquiterpenoids, especially those with guaiane-type, such as guaianediol (**1**) and alismol (**2**), and aromadendrane-type skeleton such as spathulenol (**3**) indicate their potential for further exploration. Furthermore, research on the structure-activity relationship of sesquiterpenoids offers

Fig 4. The structure-activity relationship of sesquiterpenoids **1**–**3**

excellent opportunities to optimize these compounds for drug discovery and therapeutic applications.

■ **CONCLUSION**

The phytochemical study of D. amooroides stem barks resulted in 3 sesquiterpenoid compounds, which were identified as the known compounds guaianediol (**1**), alismol (**2**), and spathulenol (**3**). The structure elucidation was performed based on an extensive spectroscopy method and a comparison with 1D-NMR data from previously reported compounds. Compounds **2** and **3** were obtained from the n-hexane extract, while **1** was isolated from the ethyl acetate. The cytotoxic assay exhibited that **3** was the strongest against breast cancer MCF-7 cells with an IC_{50} value of 15.2 μ M, while 2 possessed weak activity with an IC_{50} value of 82.1 μ M. Compound **3** was more potent than their positive control cisplatin (IC₅₀ 53.0 μ M). The structure-activity relationship implied that the presence of an olefinic bond $[-CH_2=C-]$ at C-10/C-14 in a guaiane type, also an alteration from an isopropyl moiety in the guaiane skeleton to a cyclopropane ring in the aromadendrane core can boost its activity. In addition, according to both current and previous works, the naturally occurring guaiane- and aromadendrane-type sesquiterpenoids are worth further investigation for drug development, including anticancer, antibacterial, and inflammatory diseases.

■ **ACKNOWLEDGMENTS**

The authors are grateful to Mr. Kansi Haikal at the Center Laboratory of Universitas Padjadjaran for performing the HR-ESI-TOF-MS measurement. The authors thank Universitas Padjadjaran, Indonesia, for financially supporting this investigation through the Academic Leadership Grant (No: 1630/UN6.3.1/PT.00/2024 by Unang Supratman.

■ **CONFLICT OF INTEREST**

The authors declare that there is no conflict of interest.

■ **AUTHOR CONTRIBUTIONS**

Conceptualization, Latifah Gunawan, Al Arofatus Naini, Unang Supratman; methodology, software, validation, formal analysis, investigation, resources, data curation, writing—original draft preparation, writing review and editing, Latifah Gunawan, Hidayat Nurul Mustofa, Al Arofatus Naini, Desi Harneti, Ace Tatang Hidayat, Nurlelasari, Sofa Fajriah, Khalijah Awang, Mohamad Nurul Azmi; visualization, supervision, project administration, Rani Maharani, Tri Mayanti, Khalijah Awang, Mohamad Nurul Azmi, Unang Supratman. All authors have read and agreed to the published version of the manuscript.

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