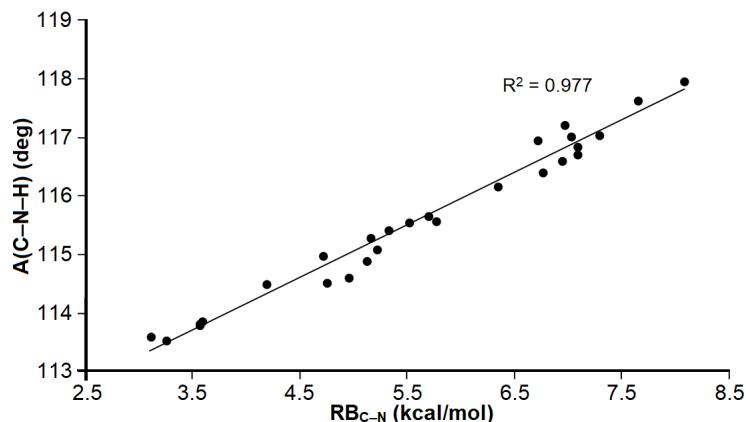
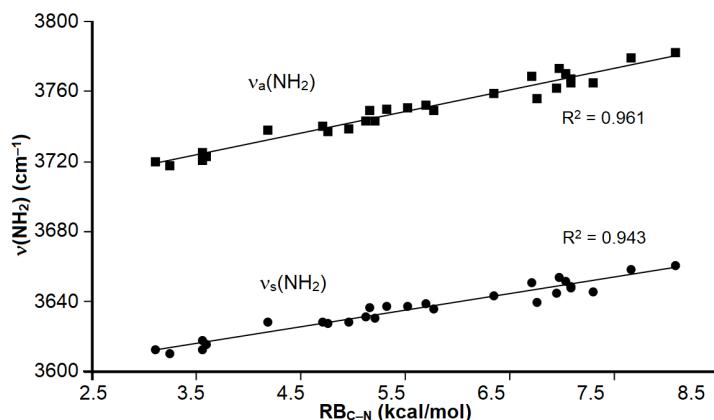


### Supplementary Data

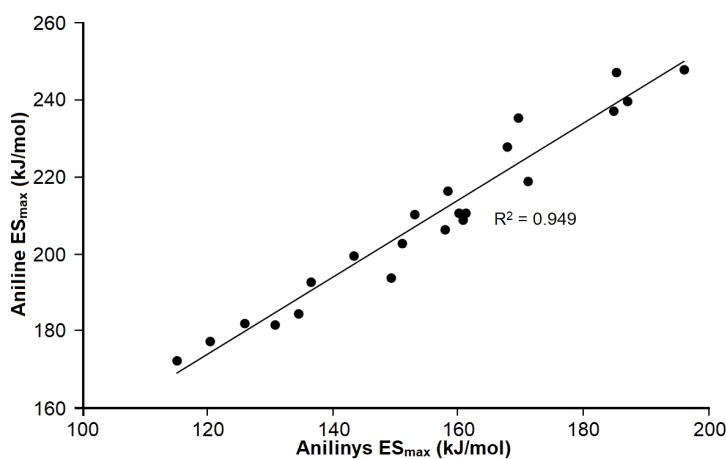
This supplementary data is a part of paper entitled “Rotational Barrier and Bond Dissociation Energy and Enthalpy: Computational Study of the Substituent Effects in Para-Substituted Anilines and Phenols”.



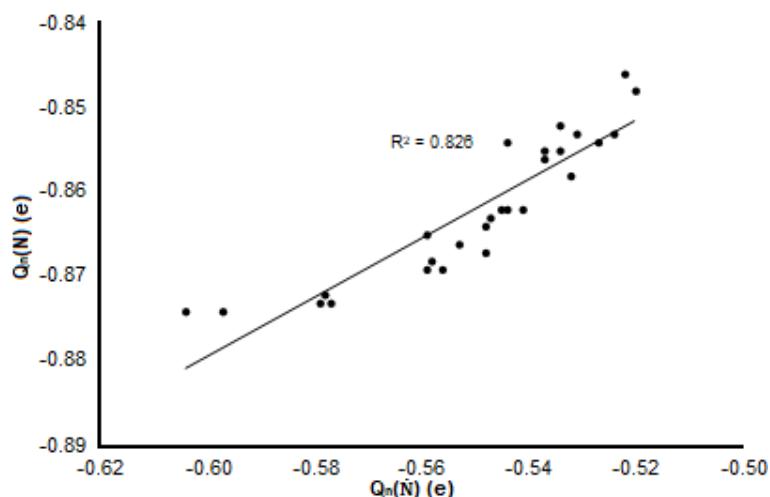
**Fig S1.** Phenyl–N–H bond angle vs rotational barrier around pheny–NH<sub>2</sub> bond in *para*-substituted anilines



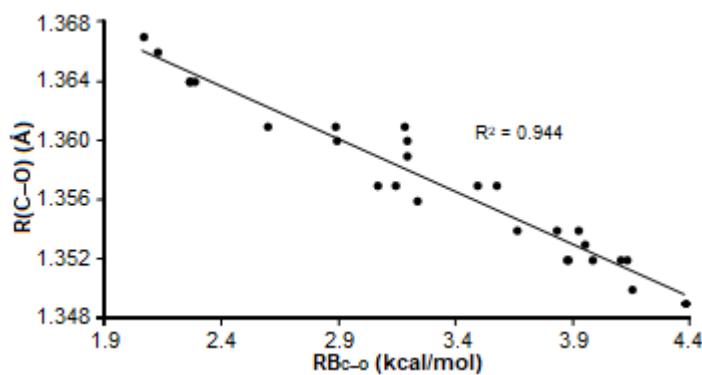
**Fig S2.** Symmetric and asymmetric NH<sub>2</sub> stretching frequencies vs rotational barrier around pheny–NH<sub>2</sub> bond in *para*-substituted anilines



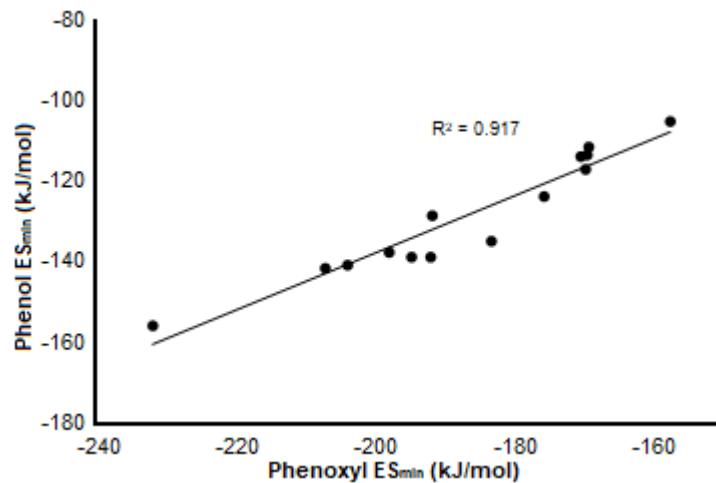
**Fig S3.** Maximum electrostatic potential around the amino H atom in *para*-substituted anilinic compounds vs that in *para*-substituted anilinyl radicals



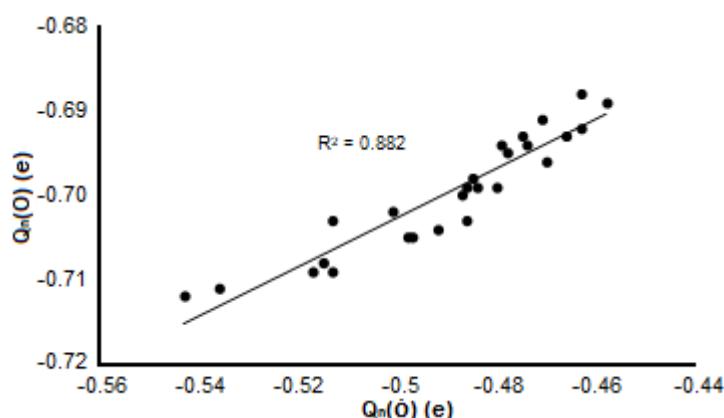
**Fig S4.** Natural partial charge on amino N atom of *para*-substituted anilines vs that in *para*-substituted anilinyl radicals



**Fig S5.** Phenyl-OH bond distance vs rotational barrier around phenyl-OH bond in *para*-substituted phenols



**Fig S6.** Minimum electrostatic potential around the hydroxyl O atom in *para*-substituted phenolic compounds vs that in *para*-substituted phenoxy radicals



**Fig S7.** Natural partial charge on hydroxyl O atom of *para*-substituted phenols vs that in *para*-substituted phenoxy radicals

**Table S1.** Symmetric and asymmetric NH<sub>2</sub> stretching frequencies,  $\nu_s(\text{NH}_2)$  and  $\nu_a(\text{NH}_2)$ , respectively, for neutral anilinic compounds and the  $\dot{\text{N}}\text{H}$  stretching frequency,  $\nu(\dot{\text{N}}\text{H})$ , of the corresponding anilinyl radicals

Substituent	Anilines		Anilinyls
	$\nu_s(\text{NH}_2)$ (cm <sup>-1</sup> )	$\nu_a(\text{NH}_2)$ (cm <sup>-1</sup> )	$\nu(\dot{\text{N}}\text{H})$
COCl	3661	3782	3487
NO <sub>2</sub>	3659	3779	3493
CHO	3646	3765	3492
COOH	3648	3765	3491
SiF <sub>3</sub>	3649	3767	3489
CN	3652	3770	3493
SO <sub>2</sub> CH <sub>3</sub>	3654	3773	3491
COCH <sub>3</sub>	3645	3762	3491
CO <sub>2</sub> CH <sub>3</sub>	3640	3756	3486
SCN	3651	3769	3494
CF <sub>3</sub>	3644	3759	3490
SCH <sub>3</sub>	3636	3749	3489
SH	3639	3752	3489
I	3638	3751	3491
Br	3638	3750	3491
Phenyl	3631	3743	3483
Cl	3637	3749	3491
H	3632	3743	3484
C(CH <sub>3</sub> ) <sub>3</sub>	3629	3739	3485
CH <sub>3</sub>	3628	3737	3486
OPhenyl	3629	3740	3490
F	3629	3738	3491
OCH <sub>3</sub>	3616	3723	3490
OH	3618	3725	3490
OCH <sub>2</sub> CH <sub>3</sub>	3613	3721	3491
N(CH <sub>3</sub> ) <sub>2</sub>	3611	3718	3488
NH <sub>2</sub>	3613	3720	3489