

### Supplementary Data

This supplementary data is a part of the paper entitled “Synthesis and Certification of Lanthanum Oxide Extracted from Monazite Sand”.

**Supplement 1 (S-1).** Homogeneity test of macroelement (La concentration) in the ten lanthanum oxide subsamples

La-Oxide Sub-sample	Test Results of La (%)		Xt	Xt-Xr	$(Xt-Xr)^2$	Wt	$Wt^2$
	A	B					
1	77.016	77.116	77.0660	-0.9867	0.9736	0.1000	0.0100
2	78.706	78.082	78.3940	0.3413	0.1165	0.6240	0.3894
3	76.636	75.622	76.1290	-1.9237	3.7006	1.0140	1.0282
4	76.890	75.958	76.4240	-1.6287	2.6527	-0.9320	0.8686
5	81.372	78.846	80.1090	2.0563	4.2284	2.5260	6.3807
6	79.810	78.232	79.0210	0.9683	0.9376	1.5780	2.4901
7	80.316	79.118	79.7170	1.6643	2.7699	-1.1980	1.4352
8	79.676	77.718	78.6970	0.6443	0.4151	1.9580	3.8338
9	76.220	77.738	76.9790	-1.0737	1.1528	1.5180	2.3043
10	79.070	76.912	77.9910	-0.0617	0.0038	2.1580	4.6569
		Xr	<b>78.0527</b>				
			Total		16.9510		23.3972
				Sx	0.9417	Sw	0.5849
				Sx <sup>2</sup>	0.8868	Sw <sup>2</sup>	0.3421
					Sw <sup>2</sup> /2		0.1711
					Sx <sup>2</sup> - (Sw <sup>2</sup> /2)		0.7158
					Ss		<b>0.3579</b>

In Supplement 1, La concentration measured of 78.0527% or 0.780527.  $\log 0.780527 = -0.106$ , then  $1 - 0.5 \log 0.780527 = -1.0538$ . The value of  $\sigma = 2.0760$ , so the value of  $0.3 \sigma = 0.6228$ . The value of Ss = 0.3579. The concentration of La has been homogeneous, because of Ss < 0.3  $\sigma$ .

**Supplement 2 (S-2).** Homogeneity test of microelement (Sm concentration) in the ten lanthanum oxide subsamples

La-Oxide Sub-sample	Test Results of Sm (%)		Xt	Xt-Xr	$(Xt-Xr)^2$	Wt	$Wt^2$
	A	B					
1	0.116	0.110	0.1130	-0.01015	0.0001	-0.0060	$3.60 \times 10^{-5}$
2	0.104	0.116	0.1100	-0.01315	0.0002	-0.0120	$1.44 \times 10^{-4}$
3	0.128	0.126	0.1270	0.00385	$1.5 \times 10^{-5}$	0.0020	$4.00 \times 10^{-6}$
4	0.125	0.119	0.1220	-0.00115	$1.3 \times 10^{-6}$	-0.0060	$3.60 \times 10^{-5}$
5	0.113	0.125	0.1190	-0.00415	$1.7 \times 10^{-5}$	-0.0120	$1.44 \times 10^{-4}$
6	0.137	0.136	0.1365	0.01335	0.0002	0.0010	$1.00 \times 10^{-6}$
7	0.132	0.129	0.1305	0.00735	$5.4 \times 10^{-5}$	-0.0030	$9.00 \times 10^{-6}$
8	0.145	0.132	0.1385	0.01535	0.0002	0.0130	$1.69 \times 10^{-4}$
9	0.119	0.118	0.1185	-0.00465	$2.2 \times 10^{-5}$	-0.0010	$1.00 \times 10^{-6}$
10	0.117	0.116	0.1165	-0.00665	$4.4 \times 10^{-5}$	0.0010	$1.00 \times 10^{-6}$
		Xr	<b>0.12315</b>				
			Total		0.0008		$5.45 \times 10^{-4}$
				Sx	$4.7 \times 10^{-5}$	Sw	$1.36 \times 10^{-5}$
				Sx <sup>2</sup>	$2.2 \times 10^{-9}$	Sw <sup>2</sup>	$1.86 \times 10^{-10}$
						Sw <sup>2</sup> /2	$9.28 \times 10^{-11}$
						Sx <sup>2</sup> - (Sw <sup>2</sup> /2)	$2.10 \times 10^{-9}$
						Ss	<b><math>1.05 \times 10^{-9}</math></b>

In Supplement 2, the Sm concentration measured of 0.123% or 0.00123.  $\log 0.00123 = -2.9096$ . The value of  $\sigma = 5.4823$ , so the value of  $0.3 \sigma = 1.6447$ . The value of  $S_s = 1.05 \times 10^{-9}$ . The concentration of Sm has been homogenous, because of  $S_s < 0.3 \sigma$ .

**Supplement 3 (S-3).** Homogeneity test of microelement (Y concentration) in the ten lanthanum oxide subsamples

La-Oxide Sub-sample	Test results of Y (%)		Xt	Xt-Xr	$(Xt-Xr)^2$	Wt	Wt <sup>2</sup>
	A	B					
1.	0.101	0.099	0.1000	-0.00165	$2.7 \times 10^{-6}$	-0.002	$4.0 \times 10^{-6}$
2.	0.096	0.101	0.0985	-0.00315	$9.9 \times 10^{-6}$	-0.005	$2.5 \times 10^{-4}$
3.	0.106	0.103	0.1045	0.00285	$8.1 \times 10^{-6}$	0.003	$9.0 \times 10^{-6}$
4.	0.100	0.103	0.1015	-0.00015	$2.2 \times 10^{-8}$	0.003	$9.0 \times 10^{-6}$
5.	0.105	0.125	0.1150	0.01335	$1.8 \times 10^{-4}$	-0.020	$4.0 \times 10^{-4}$
6.	0.095	0.100	0.0975	-0.00415	$1.7 \times 10^{-5}$	-0.005	$2.5 \times 10^{-5}$
7.	0.099	0.097	0.0980	-0.00365	$1.3 \times 10^{-5}$	-0.002	$4.0 \times 10^{-6}$
8.	0.094	0.099	0.0965	-0.00515	$2.6 \times 10^{-5}$	-0.005	$2.5 \times 10^{-5}$
9.	0.104	0.103	0.1035	0.00185	$3.4 \times 10^{-6}$	-0.001	$1.0 \times 10^{-6}$
10.	0.102	0.101	0.1015	-0.00015	$2.2 \times 10^{-8}$	0.001	$1.0 \times 10^{-6}$
	Xr	<b>0.10165</b>					
		Total			$2.6 \times 10^{-4}$		$5.0 \times 10^{-4}$
			Sx		$1.4 \times 10^{-5}$	Sw	$1.2 \times 10^{-5}$
			Sx <sup>2</sup>		$2.1 \times 10^{-10}$	Sw <sup>2</sup>	$1.6 \times 10^{-10}$
						Sw <sup>2</sup> /2	$7.9 \times 10^{-11}$
						Sx <sup>2</sup> - (Sw <sup>2</sup> /2)	$1.3 \times 10^{-10}$
						Ss	$6.4 \times 10^{-11}$

In Supplement 3, the Y concentration measured of 0.10165% or 0.0010165. The value of  $\sigma = 5.6429$ , so the value of  $0.3 \sigma = 1.6929$ . The value of  $S_s = 6.4 \times 10^{-11}$ . The concentration of Y has been homogenous, because of  $S_s < 0.3 \sigma$ .

**Supplement 4 (S-4).** Stability test of macroelement (La concentration) in the three lanthanum oxide subsamples

La-Oxide Sub-sample	The concentration of La after 6 months (%)		Y <sub>T</sub> (%)	
	Y <sub>A</sub>	Y <sub>B</sub>		
1	77.066	79.980	78.957	
4	76.424	79.160	77.792	
9	76.979	78.330	77.655	
	Yr (%)		<b>78.135</b>	

In Supplement 1 (S-1) the concentration of La was obtained Xr = 78.0527% and in supplement 4 (S-4) the stability test data of La was obtained Yr = 78.1350%, the value of  $|Xr - Yr| = 0.0828$ , the value of  $0.3 \sigma = 0.6228$ , so the concentration of La in lanthanum oxide is stable, because of  $|Xr - Yr| \leq 0.3 \sigma$  namely  $0.0828 < 0.6228$ .

**Supplement 5 (S-5).** Stability test of microelement (Sm concentration) in the three lanthanum oxide subsamples

La-oxide Sub-sample	The concentration of Sm after 6 months (%)		Y <sub>T</sub> (%)	
	Y <sub>A</sub>	Y <sub>B</sub>		
1	0.113	0.127	0.120	
2	0.110	0.119	0.115	
7	0.131	0.126	0.129	
	Yr (%)		<b>0.121</b>	

In Supplement 2 (S-2) the Sm concentration was obtained,  $X_r = 0.123\%$  and on the stability test data in Supplement 5 (S-5) the Sm concentration was obtained  $Y_r = 0.121\%$ , the value of  $|X_r - Y_r| = 0.002$ , and the value of  $0.3 \sigma = 1.6447$ , so the concentration of Sm in lanthanum oxide was stable because of  $|X_r - Y_r| \leq 0.3 \sigma$  namely  $0.002 < 1.6447$ .

**Supplement 6 (S-6).** Stability test of microelement (Y concentration) in the three lanthanum oxide subsamples

La-oxide Sub-samples	The concentration of Y after 6 months (%)		$Y_T$ (%)
	$Y_A$	$Y_B$	
3	0.105	0.101	0.103
7	0.098	0.100	0.099
4	0.101	0.099	0.100
	<b><math>Y_r</math> (%)</b>		<b>0.1007</b>

In the Supplement 3 (S-3) the Y concentration was obtained,  $X_r = 0.10165\%$  and on the stability test data, in Supplement 6 (S-6) the Y concentration was obtained  $Y_r = 0.1007\%$ , the value of  $|X_r - Y_r| = 0.0095$ , and the value of  $0.3 \sigma = 1.6919$ , so the concentration of Y in lanthanum oxide was stable because of  $|X_r - Y_r| \leq 0.3 \sigma$  namely  $0.0095 < 1.6919$ .