THE ANTIBACTERIAL COMPOUND COLLISMICIN A DERIVED FROM MARINE Streptomyces sp Q-629K

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ABSTRACT

In our course of screening for secondary metabolite derived from marine bacterial, we isolate the antimicrobial compound collysmicin A from the ethyl acetate extract of Streptomyces sp Q-629K. Separation of this compound was carried out by silica gel open column chromatography. Purification of an active compound was done using HPLC C18 with acetonitril-water system. Determination of chemical structure was done by ¹H, ¹³C NMR and LC-MS analysis. Collysmicin A was contained in fraction 3, fraction 7.2 and fraction 8.7. The antimicrobial assayed from purified compound Fr.8.7 gave diameter inhibition approximately 13 mm against S. aureus and 12 mm against B. subtilis.

Keywords: antimicrobial, collismycin A and marine Streptomyces sp.

INTRODUCTION

The discovery of penicillin in 1929 heralded the era of antibiotics and realization that microorganism are a rich source of clinically useful natural products. Since that time, between 30,000 and 50,000 natural products have been discovered from microorganisms. Of these substances, more than 10,000 are biologically active and more than 8,000 are antibiotics [1,2]. Due to this potential, over 100 microbial products are in use today as antibiotics, antitumor agents, and agrichemicals. Most antibiotics of microbial origin come from terrestrial bacteria belonging to one taxonomic group, the order Actinomycetales. Unfortunately, the rate of discovering novel metabolites from terrestrial actinomycetes is decreasing and that new sources of bioactive natural products must be explored. One such resource can be found in marine microorganisms.

Two serious problems which prevent development of drugs from marine natural products were the majority of bioactive compounds have complex structures and the limiting of samples to let on chemical synthesis. Many researchers believed that these bioactive compounds are actually produced by 'symbiotic' microorganisms. Yet bacteria and fungi which produce highly promising metabolites have not been isolated from parent organisms. However Faulkner and coworkers have obtained fascinating results with sponge or bacteria association. If such symbiotic bacteria are culturable, we can expect a more promising future in 'Drugs from the Sea'. Strictly speaking, the discovery of new structures is very important. Marine microbes are relatively new targets for such research. Unculturable microbes should be interesting to be explored. [3].

Marine bacteria are the best studied chemically of any of the marine microorganisms. This field is growing rapidly and there are frequent additions to growing literature. The first documented identification of a bioactive marine bacterial metabolite was the highly brominated pyrrole antibiotic isolated by Burkholder and co-workers from a bacterium obtained from surface of Caribbean sea grass Thalassia. [4] The Tokyo group at the Institute of Microbial Chemistry has been an innovator in search for new metabolites from marine actinomycetes. An isolate identified as Chainia purpurgena SS-228, found in mud samples obtained from Sagami Bay, was found to produce the antibiotic benzanthraquinone [5,6]. The antibiotic showed selective inhibition of Gram-positive bacteria, with MIC values of between 1 and 2 µg/ml, and was an active antitumor agent in vivo against Ehrlich carcinoma in mice, showing significant life span extension at low doses. Most interestingly, the production of this antibiotic was only observed in selected seawater media containing the unique Japanese seaweed product "Kobu cha." Kobu Cha is a dried and pulverized powder produced in Japan from the brown seaweed Laminaria. The observation clearly shows that marine microorganisms have specific nutrients adaptations which relate to their natural nutrient sources in the ocean. The other bioactive compounds that isolated by Tokyo group were aplasmomycins A-C. This compounds produced by a marine actinomycete identified as Streptomyces griseus SS-20 propagating by media containing Kobu Cha under conditions (27 °C and very low nutrient) [7-10].

Several extracts of MBI, Japan bacterium collection have been screened for biologically active compounds. The actives bacterial extract will continue

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to investigate on useful substance for drug or agrichemical. This paper was conducted to investigate antibiotics target from marine bacterial *Streptomyces sp.* Q-629K.

EXPERIMENTAL SECTION

Material

The bacterial strain of *Streptomyces* sp. Q-629K was obtained from the MBI culture collection. Media culture of KBG containing Glucose 0.3%, CaCO₃ 0.2%, Bouillon 0.1% and tangle weed powder was purchase from Wako pure chemicals. Organic solvent, ethyl acetate, chloroform, methanol and acetonitrile were purchase from Merck. Antimicrobial assays was conduct against *Staphylococcus aureus* IFO 12732 and *Bacillus subtilis* IFO 3134. Antibiotic standard Penicillin G (Sigma Co. Ltd). were used as positive control.

Instruments

Instruments that used in this work are Nano space S1-2 HPLC (Shiseido) with PDA (Diode Array) detector and column X Terra C18, LC-MS (PDA detector) LCQ Advantage ion trap mass spectrometer (Thermo Electron Co.Ltd), NMR 500MHz NMR (Varian Co. Ltd.) for structural determination. Autoclave, clean bench, Shimadzu UV/Vis spectrometer, shaker incubator were used for microbial preparation.

Procedure

Cultivation

Propagating of semi-large scale strain *Streptomyces* sp. Q629K was done in the 3 x 1 L KBG media containing Glucose 0.3%, CaCO₃ 0.2%, Bouillon 0.1% and tangle weed powder (Kobu cha) diluted by 50% sea water. Cultivation was carried on the rotary shaker incubator during 6 days, at 30 °C. Measuring pH during cultivation was done to prevent an extreme acidity as the effect of decomposing carbon source.

Antibacterial assays

Antibacterial assays against *S. aureus* and *B. subtilis* was done using agar diffusion method. All about 15 μ L bacterial extracts, fractions and isolated compounds were applied into paper disc (diameter 6 mm), then put on the agar surface of the bacterial culture. After one day incubation at 30 °C, the diameter of inhibition was determined. The same procedure was applied to the antibiotic Penicillin G as a positive control.

Isolation of active compound

After 6 days cultivation, the bacterial culture was centrifuged for separating cell and secondary metabolite. The supernatant was extracted using ethyl acetate to get organic phase substances. The pellet was extracted with ethanol 70%. All of extract were evaporated and applied for bioassay.

The active ethyl acetate phase was led to the separation by column silica gel developing using chloroform-acetonitrile by gradient method. About 8 fractions was collected by this open column separation. All of fractions were applied to the antimicrobial assays. There are 3 active fractions (Fr.3, Fr.7 and Fr.8), but only fraction Fr.7 and Fr.8 could be continued to the purification processes.

The potent open column fractions (Fr.7 and 8) was chromatographed using HPLC/PDA detector by column X Terra C18, 4.6 x 250 mm with mobile phase acetonitrile-water containing 0.1% TFA (Triflouroacetic Acid).

Structural Determination of the potent compound

Structural determination was done toward all of active antimicrobial compounds. The LC-MS (PDA detector) with Inertsil ODS-2 column diameter 1.5 x 250 mm, eluted with 70% Water containing 0,1% TFA and 30% methanol containing 0,1% TFA and ESI mode for the MS analysis was applied to detect the molecular weight of potent compound.

Analysis of ¹H, ¹³C, COSY and HMBC were done using 500 MHz NMR (Varian co.ltd).

RESULT AND DISCUSSION

Cultivation of Streptomyces sp Q629K

The pH on zero day cultivation of *Streptomyces* sp was 7.11, and after harvesting was 7.10 for *Streptomyces* sp. These value indicated that the bacterial growth on pH neutral (around 7.1 - 7.5). Thus the reason why in media containing glucose must be added calcium carbonate for prevent decreasing pH because of carbohydrate hydrolysis.

Antimicrobial Assay

The antibacterial activities was showed by clear hallow around paper disc on agar media. Diameter inhibition determined the strongest activity after cultivating overnight at 30 °C. Table 1 demonstrate activities antibacterial of extracts and fractions of this bacterial as mention above.

Tabel 1. Anti-gram positive bacterial assay ofStreptomyces sp Q-629K extract.

No	Sample	Diameter Inhibition (mm) against		
		Staphylococcus Bacillus subtilis		
aureus				
1 C	-629K. EtOAC	22	25	
2 C	2-629K. EtOH	-	-	
3 C	2-629K. water	-	-	

Tabel 2. Anti-bacterial assay of column fractions <i>Streptomyces</i> sp Q-629K.				
Sample	Mobile phase	Diameter inhibition (mm) against		
Coulomn fractions		S. aureus	B. subtilis	
Q-629K-EtOAc Fr.1	CHCl ₃ 100%	7*	8*	
Q-629K-EtOAc Fr.2	CHCl₃ 100%	-	9*	
Q-629K-EtOAc Fr.3	(CHCl ₃ /CH ₃ CN) 11/1	17	13	
Q629K-EtOAc Fr.4	(CHCl ₃ /CH ₃ CN) 4/1	16	13	
Q629K-EtOAc Fr.5	(CHCl ₃ /CH ₃ CN) 4/1	-	-	
Q629K-EtOAc Fr.6	(CHCl ₃ /CH ₃ CN) 1/1	-	9	
Q629K-EtOAc Fr.7	(CH ₃ CN) 100%	15	15	
Q629K-EtOAc Fr.8	(MeOH) 100%	28	28	

Tabel 2. Anti-bacterial assay of column fractions Streptomyces sp Q-629K.

* the hallow was not clear

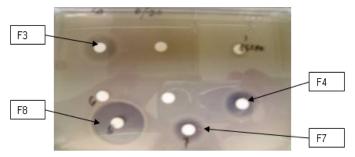


Fig 1. Inhibition effect against *S. aureus* by column fractions of ethyl acetate extract Q-629K strain

Table 3. Anti-gram positive bacterial assay by	HPLC
fractions of Streptomyces sp (Q-629K)	

No	S	ample			nhibition (mm)
				ag	gainst
				Staphylococcu	ıs Bacillus
				aureus	subtilis
1	(Q-629K) F	r.7		-	-
2	,,	HPLC	Fr.7.1	-	-
2 3	,,	,,	Fr.7.2	7*	9*
4	,,	,,	Fr.7.3	-	-
4 5 6	,,	,,	Fr.7.4	-	-
6	,,	,,	Fr.7.5	-	-
7	,,	,,	Fr.7.6	-	-
8	,,	Fr.8		23	24
9	,,	HPLC	Fr.8.1	-	-
10	,,	,,	Fr.8.2	8	12
11	,,	,,	Fr.8.3	10*	10*
12	,,	,,	Fr.8.4	-	-
13	,,	,,	Fr.8.5	8	14
14	,,	,,	Fr.8.6	-	-
15	,,	,,	Fr. 8.7	13	12
16	,,	,,	Fr.8.8	-	-
17	,,	,,	Fr.8.9	-	-
18	,,	,,	Fr.8.10	-	-
19	,,	,,	Fr.8.11	-	-
20	,,	,,	Fr.8.12	-	-
21	Penicillin G			17.5	28

The data above present that the ethyl acetate extract of this bacteria metabolite was the potent antibacterial activity compare with ethanol and water extract. We can consider that the active compound has intermediate polarity. Base from this data then we continued for separation process of ethyl acetate extract using open column method. The table 2 showed the antibacterial activity of separated fraction using silica gel open column with chloroform and acetonitrile solvent system.

The data of anti-bacterial assays from the open column fraction showed that there are 4 potentials fraction that give positive result that inhibit *S. aureus* and *B. subtilis*. The high potential of antibiotics source appeared in this strains. The various level of activity against *S. aureus* and *B. subtilis* showed by the separated fraction of open column of this extract. The potent fraction with the strong activity was indicated by fraction 3, 4, 7 and 8. The figure 1 was present the clear zone inhibition of Staphylococcus aureus growth by these column fraction.

Otherwise the fraction 3 and 4 have strong activity, we could't continued to the later purification due to the small quantity of both fraction. We decided to purify the fraction 7 and 8. The reverse phase HPLC method was applied to the fraction 7 and 8. The table 3 showed the result of antibacterial assay by HPLC column of both fraction.

HPLC fractions of *Streptomyces* sp Q-629K strain produced active fractions Fr. 7.2, Fr. 8.2, Fr.8.3, Fr. 8.5 and Fr. 8.7. Even though very weak there is only one fraction (Fr. 7.2) from purification fraction 7 that gave a positive result against both of the pathogenic bacteria. The most potent fraction from purification fraction 8 was fraction Fr. 8.7, otherwise this activity was lower than a positive control penicillin G.

Result of Separation Active Substance

Separation active compounds of Streptomyces sp (Q-629K)

The first chromatographic separation of the active ethyl acetate phase by this strain was going on the open column silica gel using chloroform-acetonitrile. This separation let to 8 fraction with gave 4 potent active fraction as Fr.3, Fr.4, Fr.7 and Fr.8. As a mention before, the available purification was done to the fraction 7 and 8. The purification of both fractions were carried on HPLC C18 column with water-acetoni

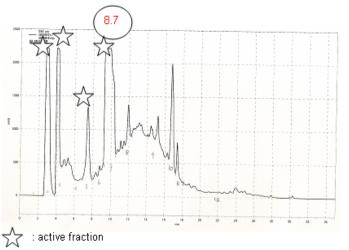


Fig 3. HPLC chromatogram of *Streptomyces* sp (Q-629K) Fr.8

Table 4. The chemical shif of ${}^{13}C$ and ${}^{1}H$ NMR of Collismycin in CDCl₃

Position	δ_{C}		δ_{H} (multiplicity ,J=Hz)
1	168.75	С	
2	158.21	С	
3	156.45	С	
4	153.96	С	
5	149.85	СН	8.63(d, 5.0)
6	148.70	CH=NOH	8.924(s)
7	138.52	СН	7.92 (t ,8.0)
8	125.61	СН	7,44(dd,8.0, 5.0)
9	123.42	С	
10	123.27	СН	8.53(d, 8.0)
11	104.86	СН	8.014 (s)
12	56.86	CH₃	4.099 (s)
13	18.34	CH₃	2.376 (s)
NOH			10.19 (br s)
			10.19 (b) 3)



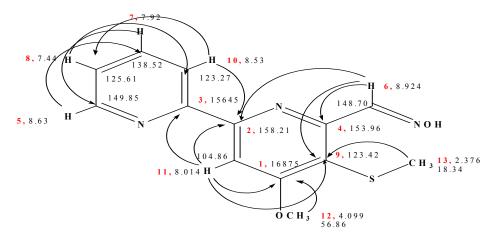


Fig 3. The structure of Collismycin A with HMBC correlation

trile containing 0.1% TFA by gradient system. Separation using HPLC method of open column fraction Fr.7 gave 6 HPLC fraction (Fr.7.1...Fr7.6). The twelve fractions (Fr8.1....Fr8.12) were collected from HPLC separation of open column 8 The HPLC chromatogram profile of open column 8 separation could be seen in the figure 2.

There are four potent fractions Fr.8.2, Fr.8.3, Fr8.5 and Fr.8.7 that appeared on the retention time 3, 4, 7 and 10 min. The available second purification was done to the fraction Fr.8.7. This fraction was chosen to the structural determination analysis.

Chemical structural determination

As a mention before, open column chromatographic separation of ethyl acetate phase of Q-629K strain led to the active fraction Fr.3, Fr.4, Fr.7 and Fr8. Chemical structural determination of Fr.3, Fr.7.2, and Fr.8.7 gave an active compound collismycin A. The detail of chemical structural analysis would be explained in the next paragraph.

The LC-MS data of Q-629K Fr.3, Fr.7.2, Fr.8.7. gave $[M+H]^{+}$ 276.2 at retention time 7.6 min. Elucidation structure of these active fractions was carried out by proton and carbon NMR analysis. Proton and carbon chemical shift could be shown in the table 4.

The ¹H NMR spectra showed there were aromatically proton at position 5 δ_H 8.63; position 7, δ_H 7.92; position 8, δ_H 7,44, position 10, δ_H 8.53 and position 11, δ_H 8.014 respectively. The duplet peaks at position 6, δ_H 8.924 considered to the proton that attached carbon – nitrogen bond (HC-N~). The singlet peak at position 12 corresponded to the proton on ~O-CH₃ bond. The last position at shift, δ_H 2.376 indicated proton methyl that attach sulphuric such as ~S-CH₃.

Study on several types of carbon NMR methods such as two dimensional C-H HSQC, HMBC and NOE deduced that the reasonable structure of isolated compound Fr.3 was, Collismycin A (Figure 3). The presence of shift C-13 at δ_C 18.34 indicated signal of ~S-CH3 bond. This data supported to the HMBC spectral signed that only one correlation between proton position 13 at shift, δ_H 2.376 with carbon C-9, δ_C 123.27.

Literature study proved that the spectral data of isolated compound was very similar with antibacterial & antitumor compound collismycin A that already isolated by Shindo, S. *et.al* [10] from *Streptomyces sp.* MQ22. This compound had been reported active against *S. aureus*, *B. subtilis*, *E. coli*, *C. albicans*, *S. cerevisiae* and *A. niger* [11]. GOMI *et. al.* [12] informed that this compound has antitumor activity and showed cytotoxic against L1210 murine leukemia cells (IC_{50} 0.08 µg/mL and 0.12 µg/mL, respectively).

CONCLUSION

Base to explanation mentioned in this paper, We could conclude that the antibacterial compound contained in ethyl acetate extract of marine bacterium *Streptomyces* sp (Q-629K) was collismycin A. In this study, this compound was inhibit *S. aureus* and *B. subtillis.*

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